

AMENDMENTS TO THE SPECIFICATION

Please replace the paragraph beginning page 14, line 7 with the following amended paragraph:

As becomes clear from Table 1, the lithium-containing complex oxides of Synthesis Examples 1 to 9 is a complex oxide within a composition range represented by General Formula: $\text{Li}_{1+x+\alpha}\text{Ni}_{(1-x-y+\delta)/2}\text{Mn}_{(1-x-y-\delta)/2}\text{M}_y\text{O}_2$ (where $0 \leq x \leq 0.15$, $-0.05 \leq x+\alpha \leq 0.2$, $0 \leq y \leq 0.4$; $-0.1 \leq \delta \leq 0.1$ (when $0 \leq y \leq 0.2$) or $-0.24 \leq \delta \leq 0.24$ (when $0.2 < y \leq 0.4$); and M is at least one element selected from the group consisting of Mg, Ti, Cr, Fe, Co, Cu, Zn, Al, Ge, Zr and Sn), in which secondary particles are formed of flocculated primary particles. Also, in the complex oxides of Synthesis Examples 1 to 6 whose primary particles have a mean particle diameter ranging from 0.3 to 3 μm and secondary particles have a mean particle diameter ranging from 5 to 20 μm , the mixture density of the formed positive electrode (at least 2.9 g/cm³) was at substantially the same level as conventionally-used LiCoO_2 of Comparative Synthesis Example 1 and achieved improved filling property. On the other hand, even within the above-noted composition range, the mixture density was found to be lower than the above and at substantially the same level as LiMn_2O_4 of Comparative Synthesis Example 2 if the primary particles or the secondary particles had a mean particle diameter beyond the scope of claims of the present invention.